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Interacts With Solvent Molecules. The Tert-butoxide Ion Is Very Bulky, Or Sterically Hindered, And Is Less Capable Of Interacting With The Solvent. The Ethoxide Ion Is Not As Sterically Hindered So It Can Accommodate More Solvent Interactions. As A Result Ethoxide Is Better Solvated And Is Therefore More Stable Than Tert-butoxide (Figure 3.7). May 24th, 2022

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C) 1,1-dibromo-2-chloro-2-fluoropropane D) 1,1-dibromo-2,2 Dichlorofluoropropane
12 Ethyne Is A(n) A) Alkane B) Alkene C) Alcohol D) Aromatic E) Alkyne 13 How
Many Alkyne Isomers Are Possible With The Molecular Formula C₆H₁₀? A) 3 Or Less
B) 4 C) 5 D) 6 E) 7 Or More 14 Alkanes Have The General Formula A) C_nH_{2n+2} B)
C_nH_{2n} C) C_nH_{2n-2} D) C_nH_n E ... Jun 13th, 2022

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